# Quizlet

# FAMOUS SCIENTISTS: LC CHEMISTRY

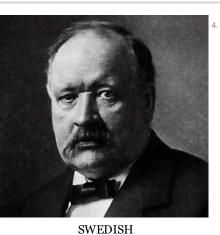
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### 1. SVANTE AUGUST ARRHENIUS

He developed a theory of acids and bases on how they form ions in solution. He also explained the fact that most reactions require added heat energy to proceed by formulating the concept of activation energy.

### 2. AMEDEO AVOGADRO

He is most noted for his contributions to molecular theory, including a law named after him. In tribute to him, the number of atoms, molecules, ions or other particles in 1 mole of a substance, 6.02214179(30)×1023, is known as a constant that also bears his name.



SWEDISH

1859 - 1927

# 4. ANTOINE HENRI BECQUEREL

Discover of radioactivity along with Marie Curie and Pierre Curie.



FRENCH





ITALIAN

1776 - 1856

# NIELS HENDRIK DAVID BOHR

Made some of the first contributions to understanding atomic structure and quantum mechanics.



DANISH

1885 - 1962

### 3. JOHANN JAKOB BALMER

Devised an empirical formula for working out the visible spectral lines (wavelength) of the hydrogen atom



### GERMAN

### 6. ROBERT BOYLE

Developed the Law (named after him) which states that for a fixed amount of an ideal gas kept at a fixed temperature, P [pressure] and V [volume] are inversely proportional (while one increases, the other decreases.



IRISH

1627 - 1691

1825 - 1898

#### 7. LOUIS DE BROGLIE

Developed the theory of electron waves and stated that any moving particle or object had an associated wave.



FRENCH

1892 - 1987

### 8. ROBERT WILHELM BUNSEN

Developed the laboratory heating apparatus named after him and with Gustav **Kirchhoff discovered** caesium (in 1860) and rubidium (in 1861).



GERMAN

1811 - 1899

### 9. ANDERS CELSIUS

He proposed the temperature scale where the boiling point of water is 100 degrees.



SWEDISH

1701 - 1744

# 10. JAMES **CHADWICK**

**English Nobel** laureate in physics awarded for his discovery of the neutron.



ENGLISH

1891 - 1974

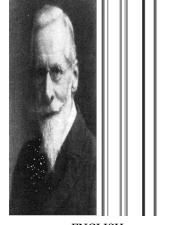


FRENCH

1746 - 1823

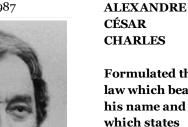


1895



ENGLISH

1832 - 1919



**CHARLES** Formulated the law which bears

11. JACQUES

his name and which states that, at constant pressure, the volume occupied by a fixed weight of gas is directly proportional to the absolute temperature

12. WILLIAM

He was a pioneer of vacuum tubes, inventing the tube, which bears his name. Wilhelm Röntgen discovered xrays with the Crookes tube in

#### 13. JOHN DALTON

He is best known for his pioneering work in the development of modern atomic theory and his research into colour blindness.



ENGLISH

1766 - 1844

### 14. DANIEL GABRIEL FAHRENHEIT

Who is best known for inventing the alcohol thermometer (1709), the mercury thermometer (1714), and for developing a temperature scale now named after him



POLISH

# 15. JOSEPH LOUIS GAY

- LUSSAC

Known mostly for two laws related to gases, and for his work on alcoholwater mixtures, which led to the degrees Gay-Lussac used to measure alcoholic beverages in many countries.



FRENCH

1778 - 1850

### 16. FRITZ HABER

Received the Nobel Prize in Chemistry in 1918 for his development for synthesizing ammonia, important for fertilizers and explosives



GERMAN

1868 - 1934

# 17. STEPHEN WILLIAM HAWKING

Key scientific works to date have included providing, with Roger Penrose, theorems regarding gravitational singularities in the framework of general relativity, and the theoretical prediction that black holes should emit radiation (Hawking radiation).

# 18. GERMAIN HENRI HESS

Developed Hess's Law that states that in a series of chemical reactions, the total energy gained or lost depends only on the initial and final states, regardless of the number or path of the steps.

19. JOSEPH JOHN

THOMPSON

spectrometer

He is credited for the

invention of the mass

discovery of the electron and of isotopes, and the



ENGLISH





SWISS

1802 - 1850



ENGLISH

1856 - 1940

### 20. JOSEPH PRIESTLEY

He is usually credited with the discovery of oxygen, having isolated it in its gaseous state.



ENGLISH

1733 - 1804

### 21. JAMES PRESCOTT JOULE

Discovered the theory of conservation of energy which led to the development of the first law of thermodynamics. The Joule is named after him.

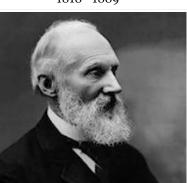


ENGLISH

1818 - 1889

# 22. WILLIAM THOMSON (LORD KELVIN)

Widely known for developing the basis of absolute zero.



NORTHERN IRISH

1824 - 1907

# 23. ANTOINE -LAURENT DE LAVOISIER

He stated the first version of the law of conservation of mass, recognized and named oxygen (1778) and hydrogen (1783), helped construct the metric system and wrote the first extensive list of elements



She was a pioneer in the field of radioactivity and the first person honored with two Nobel Prizes



FRENCH

1743 - 1794



POLISH

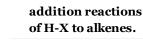
1867 - 1934



RUSSIAN

1838 - 1904

Best known for Markovnikov's rule, which he developed in 1869 to describe



25. VLADIMIR

VASILEVICH MARKOVNIKOV

# 26. DMITRI IVANOVICH MENDELEEV

Credited as being the creator of the first version of the periodic table of elements and using it to predict the properties of elements yet to be discovered.



RUSSIAN

1834 - 1907

### 27. LINUS CARL PAULING

Worked on chemical bonding and many of his contributions like hybridization and electronegativity have become part of standard chemistry textbooks. He also discovered the alpha helix and beta sheet structure of proteins.

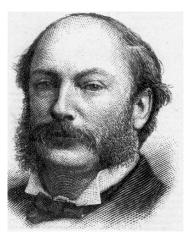


AMERICAN

1901 - 1994

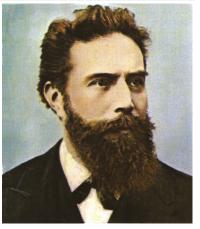
# 29. JOHN WILLIAM STRUTT (LORD RAYLEIGH)

Discovered the element argon. He also discovered the phenomenon now called Rayleigh scattering, explaining why the sky is blue.



ENGLISH

1842 - 1919

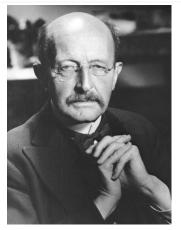


GERMAN

1845 - 1923

28. MAX KARL ERNST LUDWIG PLANCK

> Considered to be the founder of the quantum theory



GERMAN

1858 - 1947

### 31. ERNEST RUTHERFORD

Discovered the concept of radioactive half life and also differentiated and named alpha and beta radiation



NEW ZEALANDER

1871 - 1937



Produced and detected electromagnetic radiation in a wavelength range today known as xrays 32. JOHANNES DIDERIK VAN DER WAALS

Famous for his work on an equation to determine the state of matter for gases and liquids in different conditions



DUTCH

1837 - 1923

# 33. ERWIN RUDOLF JOSEF ALEXANDER SCHRÖDINGER

Achieved fame for his contributions to quantum mechanics, especially the Schrödinger equation.

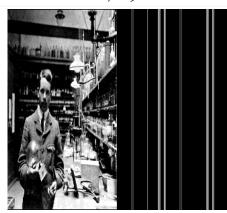


AUSTRIAN

1887 - 1961

34. HENRY MOSELEY

Provided the first experimental proof of Bohr's theory. Organised the periodic table based on atomic number rather than atomic mass and solved the Tellurium-Iodine problem.



ENGLISH

1897 - 1915